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Covalent Bonding  
Section 9.2 Naming Molecules  
In your textbook, read about how binary compounds and acids are named from their formulas.

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explaining reactions and their rates. Use  
each of the terms below just once to  
complete the...

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Early Theories of Matter In your  
textbook. read about the philosophers,  
John Dalton, and defining the atom. For  
each statement below, write true Or  
false. 1. Ancient philosophers regularly

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performed controlled experiments. 2.

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Name CHAPTER STUDY GUIDE FOR CONTENT MASTERY Section 12.2 Stoichiometric Calculations In your textbook, read about mole-to-mole conversion. Read the following passage and then solve the problems. In the equation that follows each problem, write in the space provided the mole ratio that can be used to solve the problem.

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Chemistry: Matter and Change • Chapter  
3 17 Section 3.4 Elements and  
Compounds In your textbook, read about  
elements and compounds. Circle the  
letter of the choice that best completes  
the statement or answers the question.

1. A substance that cannot be separated  
into simpler substances by physical or  
chemical

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Name \_\_\_\_\_ CHAPTER \_\_\_\_\_ Date \_\_\_\_\_ Class \_\_\_\_\_ STUDY  
GUIDE FOR CONTENT MASTERY Section  
17.2 Factors Affecting Reaction Rates In  
your textbook, read about the factors  
that affect reaction rates (reactivity,  
concentra- tion, surface, area,  
temperature, and catalysts).

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CONTENT MASTERY Name \_\_\_\_\_ Date \_\_\_\_\_ Class \_\_\_\_\_  
Study Guide for Content Mastery  
Chemistry: Matter and Change Chapter  
13 75 Section 13.2 Forces of Attraction  
In your textbook, read about forces of attraction. Answer the following questions. 1. Ionic, metallic, and covalent bonds are examples of what type of forces? 2.

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Name Cham Date CLAPTER STUDY GUIDE FOR CONTENT MASTERY SECTION 19.1 Forces within Earth, continued In your textbook read about the different

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types of faults.

## **Solved: CHAPTER 19 STUDY GUIDE FOR CONTENT MASTERY Earthqu ...**

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Chemistry: Matter and Change • Chapter  
3 17 Section 3.4 Elements and  
Compounds In your textbook, read about  
elements and compounds. Circle the  
letter of the choice that best completes  
the statement or answers the question.

1. A substance that cannot be separated  
into simpler substances by physical or  
chemical

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knowledge of content in the specific  
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